Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

T. Ravishankar,^a K. Chinnakali,^a‡ N. Arumugam,^b P. C. Srinivasan,^b Anwar Usman^c and Hoong-Kun Fun^c*

^aDepartment of Physics, Anna University, Chennai 600 025, India, ^bDepartment of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600 025, India, and ^cX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

‡ Additional correspondence author, email: kali@annauniv.edu

Correspondence e-mail: hkfun@usm.my

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.006 Å R factor = 0.044 wR factor = 0.116 Data-to-parameter ratio = 19.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved In the title compound, $C_{23}H_{19}Br_2NO_4S$, the orientation of the phenylsulfonyl substituent with respect to the indole ring system is influenced by intramolecular $C-H\cdots O$ interactions. The sulfonyl-bound phenyl ring is orthogonal to the indole ring system. In the crystal structure, $\pi-\pi$ stacking interactions involving the indole ring system link symmetry-related molecules into dimers.

Comment

Polyhalogenated indole derivatives exhibit marked antimicrobial activity against Gram-positive and Gram-negative bacteria and fungi (Piscopo, Diurno, Mazzoni & Ciaccio, 1990; Piscopo, Diurno, Mazzoni, Ciaccio & Veneruso, 1990). 5-Chloro-3-(phenylsulfonyl)indole-2-carboxamide is reported to be a highly potent non-nucleoside inhibitor of HIV-1 reverse transcriptase (Williams *et al.*, 1993). As part of our investigations on indole derivatives, we have undertaken the X-ray structure analysis of the title compound, (I).



The bond lengths and angles in (I) (Fig. 1) agree with those observed in other phenylsulfonylindoles (Ravishankar *et al.*, 2003*a,b*, 2005*a,b*; Malathy Sony *et al.*, 2005). As a result of the electron-withdrawing character of the phenylsulfonyl group, the N–Csp² bond lengths, *viz.* N1–C1 [1.420 (4) Å] and N1–C8 [1.425 (4) Å], are longer than the mean value of 1.355 (14) Å reported for N atoms with planar configurations (Allen *et al.*, 1987). Atom S1 has a distorted tetrahedral configuration, with angles O1–S1–O2 [120.61 (16)°] and N1–S1–C9 [105.02 (15)°] deviating significantly from ideal tetrahedral values.

The orientation of the phenylsulfonyl group with respect to the planar indole ring system (r.m.s. deviation = 0.011 Å) is influenced by intramolecular C-H···O interactions (Table 1) involving the sulfonyl atoms O1 and O2; these atoms deviate by 0.163 (5) and 0.029 (5) Å, respectively, from the plane of the indole ring system. As shown in Fig. 1, each of these interactions generates rings of graph-set motif S(5) or S(6) Received 15 June 2005 Accepted 6 July 2005 Online 9 July 2005 (Bernstein et al., 1995; Etter, 1990). The dihedral angle between the indole ring system and the C9–C14 phenyl ring is $89.2 (1)^{\circ}$. The 4-bromo-2,5-dimethoxybenzyl substituent is orthogonal to the indole ring system and twisted about the C15-C16 bond, as indicated by the torsion angles N1-C1-C15-C16 [89.5 (4)°] and C1-C15-C16-C17 [23.4 (5)°]. The C22-O3-C18-C17 [-8.8 (6)°] and C23-O4-C21-C20 $[-4.9 (6)^{\circ}]$ torsion angles indicate that the two methoxy substituents are essentially coplanar with the attached ring. The dihedral angle between the mean planes through the C9-C14 and C16–C21 aromatic rings is 13.2 (2)°; the centroids of these two rings are separated by 4.063 (2) Å and hence there is only a partial π - π interaction between them [C9...C17 = 3.559(5) Å, $C10 \cdot \cdot \cdot C16 = 3.426(5)$ Å, $C10 \cdot \cdot \cdot C17 =$ 3.396 (5) Å and $C11 \cdots C18 = 3.542$ (6) Å].

In the crystal structure, a π - π stacking interaction involving the indole ring system links symmetry-related molecules at (x, y, z) and $(1 - x, y, \frac{1}{2} - z)$ into a dimer; the centroid-centroid distance between the indole ring systems is 3.723 (2) Å and the perpendicular distance is 3.430 (2) Å. A view of the molecular packing down the *b* axis, illustrating the dimer formation, is shown in Fig. 2. The dimers are linked by a weak intermolecular C-H···O interaction (Table 1). A Br1···O1(*x*, 1 + *y*, *z*) short contact of 3.248 (3) Å is also observed.

Experimental

To a solution of 3-bromo-1-phenylsulfonylindol-2-ylmethanol (25 mmol) in chloroform (400 ml), a solution of 4-bromo-3methoxyacetanilide (25 mmol) in the same solvent (25 ml) was added, followed by anhydrous magnesium sulfate (10 g) and boron trifluoride etherate (2.0 ml). The resulting solution was refluxed for 3 h. Water (100 ml) was then added and the organic layer was separated. The organic layer was washed with 20% hydrochloric acid (50 ml), followed by water and saturated sodium bicarbonate solution. The solvent was removed by distillation, after drying over anhydrous sodium sulfate. The residue was chromatographed on a silica gel column (350 mesh) and eluted successively with 20, 25 and 30% ethyl acetate in hexane. The 30% ethyl acetate eluent gave the title compound, which was then crystallized from hexane–chloroform (2:1).

Crystal data

C23H19Br2NO4S
$M_r = 565.27$
Monoclinic, C2/c
a = 33.177 (2) Å
b = 9.0850 (6) Å
c = 16.1712 (10) Å
$\beta = 111.802 \ (1)^{\circ}$
$V = 4525.5 (5) \text{ Å}^3$
Z = 8
Data collection

Siemens SMART CCD areadetector diffractometer ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{min} = 0.282, T_{max} = 0.513$ 13791 measured reflections $D_x = 1.659 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 3615 reflections $\theta = 5.1-49.8^{\circ}$ $\mu = 3.71 \text{ mm}^{-1}$ T = 293 (2) KBlock, colourless $0.42 \times 0.24 \times 0.18 \text{ mm}$

5536 independent reflections 3263 reflections with $I > 2\sigma(I)$ $R_{int} = 0.041$ $\theta_{max} = 28.3^{\circ}$ $h = -41 \rightarrow 44$ $k = -12 \rightarrow 11$ $l = -21 \rightarrow 15$



Figure 1

The structure of (I), showing the atom-numbering scheme and intramolecular hydrogen bonds (dashed lines). Displacement ellipsoids are drawn at the 30% probability level.



Figure 2

The crystal packing in (I), showing the dimers; π - π nteractions are shown as dashed lines.

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.044$ $wR(F^2) = 0.116$ S = 1.015536 reflections 282 parameters H-atom parameters constrained $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0461P)^{2} + 5.2213P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.89 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.69 \text{ e } \text{\AA}^{-3}$

organic papers

Table 1Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$	
C7−H7···O2	0.93	2.32	2.906 (5)	120	
C10−H10···O1	0.93	2.52	2.899 (5)	105	
C15−H15A···O1	0.97	2.28	2.915 (5)	122	
$C12-H12\cdots O4^i$	0.93	2.60	3.394 (6)	143	

Symmetry code: (i) $x, -y, z + \frac{1}{2}$.

The H atoms were positioned geometrically and were refined as riding, with C–H distances of 0.93 (aromatic), 0.97 (methylene) and 0.96 Å (methyl), and with $U_{\rm iso}(\rm H) = 1.2 U_{\rm eq}(\rm C)$, or $1.5 U_{\rm eq}(\rm C)$ for methyl H atoms.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

HKF thanks the Malaysian Government and Universiti Sains Malaysia for research grant R&D No. 305/PFIZIK/ 610961.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1573.
- Etter, M. C. (1990). Acc. Chem. Res. 23, 120-126.
- Malathy Sony, S. M., Palani, K., Charles, P., Ponnuswamy, M. N., Sureshbabu, N., Srinivasan, P. C. & Nethaji, M. (2005). Acta Cryst. E61, 0521–0523.
- Piscopo, E., Diurno, M. V., Mazzoni, O. & Ciaccio, A. M. (1990). Boll. Soc. Ital. Biol. Sper. 66, 1181–1186.
- Piscopo, E., Diurno, M. V., Mazzoni, O., Ciaccio, A. M. & Veneruso, G. (1990). Boll. Soc. Ital. Biol. Sper. 66, 1187–1191.
- Ravishankar, T., Chinnakali, K., Arumugam, N., Srinivasan, P. C., Usman, A. & Fun, H.-K. (2003a). Acta Cryst. C59, o137–o140.
- Ravishankar, T., Chinnakali, K., Arumugam, N., Srinivasan, P. C., Usman, A. & Fun, H.-K. (2003b). Acta Cryst. E59, 01903–01906.
- Ravishankar, T., Chinnakali, K., Arumugam, N., Srinivasan, P. C., Usman, A. & Fun, H.-K. (2005a). Acta Cryst. E61, 0998–01000.
- Ravishankar, T., Chinnakali, K., Arumugam, N., Srinivasan, P. C., Usman, A. & Fun, H.-K. (2005b). Acta Cryst. E61, o1184–01186.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). SMART and SAINT. Versions 4.0. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Williams, T. M., Ciccarone, T. M., MacTough, S. C., Rooney, C. S., Balani, S. K., Condra, J. H., Emini, E. A., Goldman, M. E., Greenlee, W. J. & Kauffman, L. R. (1993). J. Med. Chem. 36, 1291–1294.